

# Ligand Binding Sites in *Escherichia coli* Inorganic Pyrophosphatase: Effects of Active Site Mutations<sup>†</sup>

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**ABSTRACT:** Type I soluble inorganic pyrophosphatases (PPases) are well characterized both structurally and mechanistically. Earlier we measured the effects of active site substitutions on pH–rate profiles for the type I PPases from both *Escherichia coli* (E-PPase) and *Saccharomyces cerevisiae* (Y-PPase). Here we extend these studies by measuring the effects of such substitutions on the more discrete steps of ligand binding to E-PPase, including (a) Mg<sup>2+</sup> and Mn<sup>2+</sup> binding in the absence of added ligand; (b) Mg<sup>2+</sup> binding in the presence of either P<sub>i</sub> or hydroxymethylbisphosphonate (HMBP), a competitive inhibitor of E-PPase; and (c) P<sub>i</sub> binding in the presence of Mn<sup>2+</sup>. The active site of a type I PPase has well-defined subsites for the binding of four divalent metal ions (M1–M4) and two phosphates (P1, P2). Our results, considered in light of pertinent results from crystallographic studies on both E-PPase and Y-PPase and parallel functional studies on Y-PPase, allow us to conclude the following: (a) residues E20, D65, D70, and K142 play key roles in the functional organization of the active site; (b) the major structural differences between the product and substrate complexes of E-PPase are concentrated in the lower half of the active site; (c) the M1 subsite is functionally isolated from the rest of the active site; and (d) the M4 subsite is an especially unconstrained part of the active site.

Soluble inorganic pyrophosphatase (EC 3.6.1.1; PPase),<sup>1</sup> which hydrolyzes inorganic pyrophosphate (PP<sub>i</sub>) to inorganic phosphate (P<sub>i</sub>), is an essential enzyme (1–3), providing a thermodynamic pull for many biosynthetic reactions (4). Two types of soluble PPases are known (5–7). Structural and functional properties of type I PPases have been analyzed quite extensively, with the best studied being those from *Escherichia coli* (E-PPase) and the cytosol of *Saccharomyces cerevisiae* (Y-PPase). A simplified kinetic model for the catalysis of type I PPases at fixed pH (7.2) is shown in Scheme 1 (8–12). Noteworthy in this scheme is the binding of two Mg<sup>2+</sup> ions in the absence of substrate and of two additional Mg<sup>2+</sup> ions in the presence of substrate.

E-PPase is a homohexamer and Y-PPase is a homodimer with subunit molecular masses of 20 kDa (13) and 32 kDa (14), respectively. We have recently identified 13 polar residues at the active site as being important for enzyme

activity through a combination of sequence comparisons (5, 15), measurements of the enzymatic activity of active site variants (16–20), and X-ray crystallography (21, 22). Results discussed in these publications have clearly demonstrated a remarkable conservation of the type I PPase active site throughout evolution, and have permitted formulation of a detailed mechanism for PPase catalysis.

Scheme 1 implies the existence of six defined ligand subsites at the enzyme active site, four for the binding of the metal ions (subsites M1, M2, M3, and M4) and two for the binding of the two product P<sub>i</sub>s (P1 and P2). PP<sub>i</sub> binding has been recently shown to overlap sites P1 and P2 (22). Crystallographic studies of E-PPase have thus far located only three of the six putative subsites—sites M1 and M2 for Mg<sup>2+</sup> binding in the absence of a phosphoryl ligand such as P<sub>i</sub> or a PP<sub>i</sub> analogue (23, 24), and site P1, seen with bound sulfate in the absence of divalent metal ion (25). More significant results have been obtained for Y-PPase, for which high-resolution structures are now available for the Mn<sub>2</sub>E, Mn<sub>2</sub>E(MnP<sub>i</sub>)<sub>2</sub>, and Mn<sub>2</sub>EFMn<sub>2</sub>PP<sub>i</sub> complexes (21, 22, 26). The latter two structures, in particular, confirm the existence of the six ligand sites. In addition to the ligand subsites at the active site, a seventh ligand subsite, with a stoichiometry of one per two subunits (27), has been detected at the trimer–trimer interface of E-PPase (23, 24).

Earlier we determined the effect of active site substitutions on pH–rate profiles for both E-PPase (16) and Y-PPase (20). Here we explore the effects of such substitutions on the more discrete steps of ligand binding to E-PPase, specifically, on (a) Mg<sup>2+</sup> and Mn<sup>2+</sup> binding to E-PPase in the absence of

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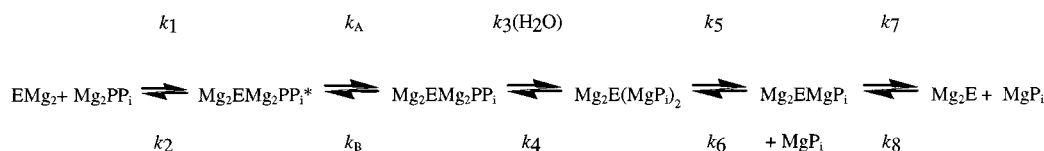
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<sup>1</sup> Abbreviations: E-PPase, *Escherichia coli* inorganic pyrophosphatase; EGTA, ethylene glycol bis(β-aminoethyl ether)-N,N,N',N'-tetraacetic acid; HMBP, hydroxymethylbisphosphonate; PPase, inorganic pyrophosphatase; PP<sub>i</sub>, inorganic pyrophosphate; P<sub>i</sub>, inorganic phosphate; Y-PPase, *Saccharomyces cerevisiae* inorganic pyrophosphatase; WT, wild-type.

Scheme 1: Minimal Scheme of Type I-PPase Catalysis<sup>a</sup>

<sup>a</sup> The sequence of catalytic events in the active site includes substrate binding to preformed enzyme complex containing two  $\text{Mg}^{2+}$  ions/subunit, isomerization of the resulting complex, P–O bond breakdown by direct attack of water, and stepwise dissociation of two phosphate molecules.

added ligand; (b)  $\text{Mg}^{2+}$  binding to E-PPase in the presence of either  $\text{P}_i$  or hydroxymethylbisphosphonate (HMBP), a competitive inhibitor of E-PPase; and (c)  $\text{P}_i$  binding in the presence of  $\text{Mn}^{2+}$ . We focus on the 13 polar active site residues involved in interactions with subsites M1–M4, P1, and P2, either directly or via an intervening water. These residues include 5 Asp, 2 Glu, 3 Lys, 1 Arg, and 2 Tyr. We employ variants either conserving charge (D/E, E/D, K/R, R/K) or conserving shape (Y/F). For a charged residue, the effects of the substitution indicate the sensitivity of ligand binding to the precise placement of the charge in question, whereas Y/F substitution examines the dependence of ligand binding on the presence of the Tyr hydroxyl. Comparison of our results with the recently determined crystal structures referred to above allows evaluation of the steric and electrostatic requirements of interactions identified in these structures.

## EXPERIMENTAL PROCEDURES

**Materials.** Wild-type E-PPase (28) and E-PPase variants free of contamination with wild-type enzyme (16) were prepared and purified as described. The disodium salt of HMBP was obtained as a gift from F. H. Ebetino (Procter and Gamble Pharmaceuticals).

**Methods.** All experiments were performed at pH 7.2 at an ionic strength of 0.1 M. Enzyme concentrations are expressed as monomer. Protein concentrations were determined by Bradford assay (Pierce). Initial rates of  $\text{PP}_i$  hydrolysis at 25 °C (29) and equilibrium dialysis experiments at  $23 \pm 2$  °C (30) were performed as described. The following buffers were employed: (a)  $\text{Mg}^{2+}$  binding—Tris-HCl buffer (0.1 M) containing 20–50  $\mu\text{M}$  EGTA; (b)  $\text{Mg}^{2+}$  binding in the presence of  $\text{P}_i$ —Tris-HCl buffer (0.04 M) containing 50  $\mu\text{M}$  EGTA, 20 mM  $\text{KH}_2\text{PO}_4$ ; (c)  $\text{Mg}^{2+}$  binding in the presence of 100  $\mu\text{M}$  HMBP—as in (a); (d)  $\text{Mn}^{2+}$  binding—Tris-HCl buffer (0.1 M); (e)  $\text{P}_i$  binding in the presence of  $\text{Mn}^{2+}$ —0.083 M TES, 0.017 M KCl, 0.1 mM  $\text{MnCl}_2$ .  $\text{Mg}^{2+}$  and  $\text{Mn}^{2+}$  concentrations were determined by atomic absorption spectrometry at 285.2 and 279.5 nm, respectively.  $\text{P}_i$  determinations were carried out by an automatic  $\text{P}_i$  analyzer (29). Enzyme-containing samples used for  $\text{P}_i$  determination were deproteinized by pretreatment with 0.2 M perchloric acid, followed by centrifugation and dilution of supernatant to an appropriate concentration.

## CALCULATIONS

Fitting of equations used in this work was performed using a program for nonlinear regression analysis (31, version 3.55) and MicroMath Scientist (version 2.01, MicroMath, Inc). Plots were constructed with SigmaPlot (version 2.01, Jandel).

**$\text{Mg}^{2+}$  or  $\text{Mn}^{2+}$  Binding.** Values of dissociation constants for  $\text{Mg}^{2+}$  or  $\text{Mn}^{2+}$  binding of up to 2.5 equiv per E-PPase

monomer were obtained by fitting equilibrium dialysis binding data to eq 1 (see ref 27), which is derived assuming that  $\text{M}^{2+}$  binding to the interface occurs independently of  $\text{M}^{2+}$  binding to the active site, and that all six monomers are identical. Here,  $n$  is the number of divalent metal ions bound per E-PPase monomer, as determined by equilibrium dialysis,  $K_1$  and  $K_2$  are the macroscopic dissociation constants for  $\text{M}^{2+}$  binding to the active site, and  $K_{\text{in}}$  is the dissociation constant for binding to each of the three identical interface sites per hexamer, i.e., one per dimer. These constants are designated  $K_{\text{g}1}$ ,  $K_{\text{g}2}$ , and  $K_{\text{g}:\text{in}}$  and  $K_{\text{n}1}$ ,  $K_{\text{n}2}$ , and  $K_{\text{n}:\text{in}}$  for  $\text{Mg}^{2+}$  and  $\text{Mn}^{2+}$ , respectively. In the equations below, E is expressed as monomer concentration and D is expressed as dimer concentration.

$$n/\text{monomer} = 0.5 \times \{[\text{M}^{2+}]/([\text{M}^{2+}] + K_{\text{in}})\} + \frac{K_2[\text{M}^{2+}] + 2[\text{M}^{2+}]^2}{(K_1K_2 + K_2[\text{M}^{2+}] + [\text{M}^{2+}]^2)} \quad (1)$$

where  $K_1 = [\text{E}][\text{M}^{2+}]/[\text{EM}^{2+}]$ ,  $K_2 = [\text{EM}^{2+}][\text{M}^{2+}]/[\text{EM}_2]$ ,  $K_{\text{in}} = [\text{M}][\text{D}]/[\text{MD}]_{\text{t}}$ ; and  $[\text{D}]_{\text{T}} = [\text{D}]_{\text{t}} + [\text{MD}]_{\text{t}}$ ,  $[\text{D}]_{\text{t}} = [\text{D}] + [\text{DM}] + [\text{DM}_2] + [\text{DM}_3] + [\text{DM}_4]$ ,  $[\text{MD}]_{\text{t}} = [\text{MD}] + [\text{MDM}] + [\text{MDM}_2] + [\text{MDM}_3] + [\text{MDM}_4]$ .

**$\text{Mg}^{2+}$  Binding in the Presence of Phosphoryl Ligands.** We assume that  $\text{Mg}^{2+}$  binds to up to 4.5 sites within the E-PPase monomer in the presence of either  $\text{P}_i$  or HMBP, 4 per active site as seen in the Y-PPase complex  $\text{Mn}_2\text{E}(\text{MnP}_i)_2$  plus 0.5 at the interface. Values of apparent dissociation constants for  $\text{Mg}^{2+}$  binding to these sites were obtained by fitting equilibrium dialysis binding data to eq 2, which was derived on the basis of the following considerations: (a)  $[\text{Mg}^{2+}]_{\text{t}}$  refers to all forms of  $\text{Mg}^{2+}$  not bound to enzyme; i.e., it includes all  $\text{Mg}_x\text{L}$  terms where L is  $\text{P}_i$  or HMBP and  $x$  is equal to 1 or 2; (b)  $K'_{\text{g}1}$ ,  $K'_{\text{g}2}$ ,  $K_{\text{g}3}$ , and  $K_{\text{g}4}$  are the macroscopic dissociation constants for the binding of the first, second, third, and fourth metal ions to the active site in the presence of phosphoryl ligand, respectively; (c) for simplicity, we may assume that  $K_{\text{g}3} = K_{\text{g}4} = K_{\text{g}3/4}$ , as both WT and variant PPase fits to the  $\text{Mg}^{2+}$  binding data in the presence of either 20 mM  $\text{P}_i$  or 100  $\mu\text{M}$  HMBP were not improved significantly when  $K_{\text{g}3}$  was allowed to be different from  $K_{\text{g}4}$ ; and (d)  $K_{\text{g}:\text{in}}$  is assumed to have the same value as in the absence of phosphoryl ligands, since these ligands do not interact with the interface site.

$$n/\text{monomer} = 0.5 \times \{[\text{M}^{2+}]_{\text{t}}/([\text{M}^{2+}]_{\text{t}} + K_{\text{in}})\} + \frac{\text{Num}}{\text{Den}} \quad (2)$$

where  $\text{Num} = K'_{\text{g}2}K_{\text{g}3/4}^2[\text{M}^{2+}] + 2K_{\text{g}3/4}^2[\text{M}^{2+}]^2 + 3K_{\text{g}3/4}[\text{M}^{2+}]^3 + 4[\text{M}^{2+}]^4$  and  $\text{Den} = K'_{\text{g}1}K'_{\text{g}2}K_{\text{g}3/4}^2 + K'_{\text{g}2}K_{\text{g}3/4}^2[\text{M}^{2+}] + K_{\text{g}3/4}^2[\text{M}^{2+}]^2 + K_{\text{g}3/4}[\text{M}^{2+}]^3 + [\text{M}^{2+}]^4$ . In the presence of  $\text{P}_i$ ,  $K'_{\text{g}1}$  is designated  $K^{\text{P}}_{\text{g}1}$ ,  $K'_{\text{g}2}$  is designated  $K^{\text{P}}_{\text{g}2}$ , and  $K'_{\text{g}3/4}$  is designated  $K^{\text{P}}_{\text{g}3/4}$ . In the presence of HMBP,  $K'_{\text{g}1}$  is designated  $K^{\text{PP}}_{\text{g}1}$ ,  $K'_{\text{g}2}$  is designated  $K^{\text{PP}}_{\text{g}2}$ , and  $K'_{\text{g}3/4}$  is designated  $K^{\text{PP}}_{\text{g}3/4}$ .

Table 1:  $\text{Mg}^{2+}$  and  $\text{Mn}^{2+}$  Binding<sup>a</sup>

E-PPase	$K_{g1}$ (mM)	$K_{g2}$ (mM)	$K_{g:in}$ (mM)	$K_{n1}$ ( $\mu\text{M}$ )	$K_{n2}$ (mM)	$K_{n:in}$ (mM)
WT	$0.076 \pm 0.002$	$6.6 \pm 0.6$	2.0	$6 \pm 1$	$0.34 \pm 0.04$	0.35
WT + 0.5 mM $\text{Mg}^{2+}$ <sup>b</sup>				$21 \pm 4$	$0.40 \pm 0.05$	0.40
WT + 5.0 mM $\text{Mg}^{2+}$ <sup>b</sup>				$135 \pm 8$	$1.4 \pm 0.1$	0.45
E20D	$0.132 \pm 0.009$	$> 50$	8.0	—	—	—
K29R	$0.072 \pm 0.002$	$3.3 \pm 0.2$	3.0	$9 \pm 3$	$0.21 \pm 0.05$	0.20
E31D	$0.067 \pm 0.002$	$2.3 \pm 0.2$	2.5	—	—	—
R43K	$0.085 \pm 0.004$	$4.0 \pm 0.5$	3.5	$7 \pm 3$	$0.19 \pm 0.05$	0.20
Y55F	$0.068 \pm 0.003$	$2.7 \pm 0.4$	2.5	—	—	—
D65E	$0.51 \pm 0.02$	$7 \pm 1$	6.5	$94 \pm 13$	$0.46 \pm 0.06$	0.30
D67E	$0.152 \pm 0.006$	$13 \pm 4$	1.5	$11 \pm 1$	$0.69 \pm 0.07$	0.20
D70E	$0.088 \pm 0.002$	$4.1 \pm 0.3$	2.5	$7 \pm 1$	$0.46 \pm 0.04$	0.45
D97E	$0.073 \pm 0.002$	$3.4 \pm 0.3$	3.5	—	—	—
D102E	$0.219 \pm 0.005$	$3.6 \pm 0.3$	3.0	$9 \pm 3$	$0.26 \pm 0.07$	0.25
K104R	$0.094 \pm 0.003$	$3.6 \pm 0.3$	3.5	—	—	—
Y141F	$0.087 \pm 0.004$	$4.4 \pm 0.5$	2.0	$10 \pm 4$	$0.16 \pm 0.05$	0.15
K142R	$0.118 \pm 0.004$	$16 \pm 5$	1.0	$9 \pm 2$	$0.21 \pm 0.04$	0.20
D97V	$0.133 \pm 0.005$	$8 \pm 2$	1.0	—	—	—
D102V	$0.87 \pm 0.03$	$27 \pm 13$	1.5	—	—	—
D26S	$0.085 \pm 0.007$	$10 \pm 5$	— <sup>c</sup>	—	—	—
K34R	$0.080 \pm 0.003$	$11 \pm 2$	2.0	—	—	—
E98V	$0.084 \pm 0.007$	$6 \pm 1$	4.0	—	—	—
H136Q	$0.057 \pm 0.005$	$2.8 \pm 0.6$	2.5	—	—	—
H140Q	$0.122 \pm 0.004$	$8 \pm 2$	1.0	—	—	—

<sup>a</sup> Enzyme concentration:  $\text{Mg}^{2+}$  binding, 0.3–1.0 mM;  $\text{Mn}^{2+}$  binding, 0.3 mM. Data for  $\text{Mg}^{2+}$  binding to WT-PPase (10) and the variants E20D (18), Y55F & K104R (19), D97E (30), and H136Q & H140Q (32) were obtained previously, and refit to eq 1, taking the interface site into account. For WT-E-PPase, considerably lower values of  $K_{g1}$  and  $K_{in}$  (although not  $K_{g2}$ ) were recently found in non-Tris-containing buffers (33) due to a Tris interaction with enzyme. However, this interaction should not greatly affect the conclusions drawn in this work, which only concern differences in ligand affinity between variants and wild-type. <sup>b</sup>  $\text{Mn}^{2+}$  binding only. <sup>c</sup> Assumed to be infinite (27).

**$P_i$  Binding.** Values of dissociation constants for  $P_i$  binding to the high-affinity site on E-PPase were fit to binding data using eq 3, where  $n$  is equal to the stoichiometry of bound  $P_i$  per E-PPase subunit,  $K_p$  is the apparent dissociation constant describing this binding, and  $[P_i]_t$  refers to all forms of  $P_i$  not bound to enzyme; i.e.,  $P_i$  plus  $\text{Mn}P_i$ .

$$n = [P_i]_t / (K_p + [P_i]_t) \quad (3)$$

**Initial Rates of  $PP_i$  Hydrolysis.** Values of  $1/v_i$  in the presence of HMBP were fit to eq 4 describing competitive inhibition.

$$1/v_i = 1/V_{\max} + K_m(1 + [\text{HMBP}]/K_i)/[\text{MgPP}_i]V_{\max} \quad (4)$$

## RESULTS

**$\text{Mg}^{2+}$  and  $\text{Mn}^{2+}$  Binding.** The three distinct sites of divalent metal ion binding to E-PPase in the absence of substrate, M1 and M2 at the active site and  $M_{in}$  at the trimer–trimer interface, bind with stoichiometries per subunit of 1.0, 1.0, and 0.5, respectively (23, 24). To determine the effects of mutation on such binding, equilibrium dialysis measurements were carried out in the absence of substrate at pH 7.2 on WT-PPase and 20 variants. Sample data for variants in which substitutions are made in residues directly binding to M1 are shown in Figure 1. The results obtained were fit to eq 1, giving the dissociation constants summarized in Table 1.

The fitting procedure required setting initial values for each of the three constants  $K_{g1}$ ,  $K_{g2}$ , and  $K_{g:in}$  and initially yielded an ambiguous result. In most cases,  $K_{g2} > K_{g1}$ ,  $K_{g:in}$  regardless of initial values. However, when initial  $K_{g1} < \text{initial } K_{g:in}$ , the fitted  $K_{g1} < \text{fitted } K_{g:in}$ , and when initial  $K_{g1} > \text{initial } K_{g:in}$ , the fitted  $K_{g1} > \text{fitted } K_{g:in}$ . For example, for WT-PPase

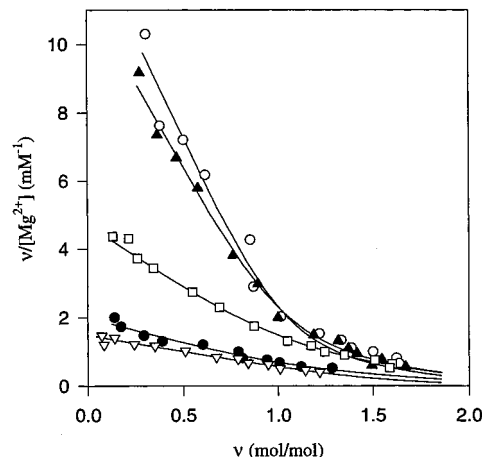


FIGURE 1: Scatchard plots of  $\text{Mg}^{2+}$  binding to WT-E-PPase and several E-PPase variants. Lines are fit to eq 1, using parameter values found in Table 1. Wild-type (O), D65E (●), D70E (▲), D102E (□), D102V (▽).

we obtained two sets of fitted values [ $K_{g1}$ ,  $K_{g2}$ ,  $K_{g:in}$ : 0.076, 6.6, 2.0 (Table 1) and 0.38, 18, 0.035, all values in millimolar], each of which fit our data quite well. This ambiguity was resolved by measuring  $\text{Mg}^{2+}$  binding to the D26S variant which lacks the interface site [Asp 26 residues from two subunits form part of the interface binding site; (23, 24, 27)]. The D26S variant shows reduced binding stoichiometry, as expected, and has  $K_{g1}$  and  $K_{g2}$  values of  $0.085 \pm 0.007$  and  $10 \pm 5$  mM, respectively. The  $K_{g1}$  value for D26S clearly shows that for WT-PPase,  $K_{g1} < K_{g:in}$ , since it is essentially identical to the value for WT-PPase obtained when initial  $K_{g1} < \text{initial } K_{g:in}$ , but is 4 times lower than the value obtained when initial  $K_{g1} > \text{initial } K_{g:in}$ . Thus, the

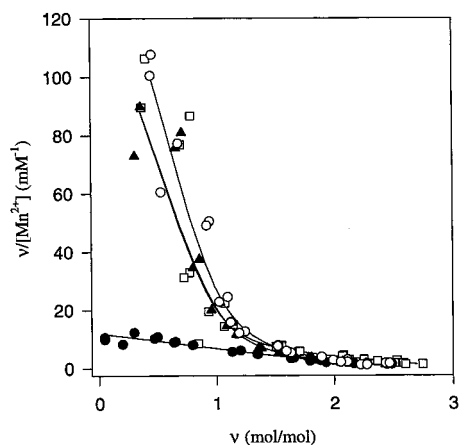


FIGURE 2: Scatchard plots of  $\text{Mn}^{2+}$  binding to WT-E-PPase and several E-PPase variants. Lines are fit to eq 1, using parameter values found in Table 1. Wild-type ( $\circ$ ), D65E ( $\bullet$ ), D70E ( $\blacktriangle$ ), D102E ( $\square$ ).

values in Table 1 were obtained setting initial  $K_{g1} < \text{initial } K_{g\text{in}}$ .

Earlier we had interpreted equilibrium dialysis studies of  $\text{Mg}^{2+}$  binding to WT-PPase and to several active site variants (18, 19, 30, 32) as reflecting binding to sites M1 and M2 only. Reinterpretation of these data according to eq 1, which takes the interface site into account, has little effect on  $K_{g1}$ , but significantly increases  $K_{g2}$ . The resulting large difference in magnitudes of  $K_{g1}$  and  $K_{g2}$ , not only for WT-PPase but also for virtually all the variants studied, has as a consequence that these macroscopic constants closely approximate microscopic dissociation constants for ordered binding to specific sites in PPase:  $K_{g1}$  to M1 and  $K_{g2}$  to M2 (see Discussion).

Besides D26S, the other 19 variants to which  $\text{Mg}^{2+}$  binding was measured include conservative (D/E, R/K, Y/F) substitutions at each of 13 conserved active site polar residues as well as 2 nonconservative variants within this group, D97V and D102V. The additional four variants include two strongly conserved polar residues not involved functionally at the active site, K34R and E98V, and two (H136Q and H140Q) at the trimer-trimer interface (32). Only variants at positions 65 and 102 show a sizable increase in  $K_{g1}$ ; the D65E and

D102E substitutions lead to 7- and 3-fold increases, respectively, whereas D102V substitution leads to an 11-fold increase. By contrast, all other substitutions, including D97V and E98V, have little effect on  $K_{g1}$ .

As  $K_{g2}$  describes binding to a weak site, substitutions leading to a further weakening are difficult to measure accurately. However, E20D and D102V substitution clearly cause large decreases in affinity ( $> 8$ -fold and  $\sim 4$ -fold, respectively), whereas most other substitutions have smaller effects, with several (e.g., E31D, Y55F, and H136Q) actually increasing affinity. Finally, whereas the great majority of substitutions have only small effects on  $K_{g\text{in}}$ , two (E20D and D65E) lead to significant increases (3–4-fold), demonstrating that some effects of active site substitution can be propagated to the interface.

$\text{Mn}^{2+}$  binding to WT-PPase and eight conservative active site variants was carried out and analyzed in an exactly analogous fashion to  $\text{Mg}^{2+}$  binding, except that a lower enzyme concentration was used because the affinities for  $\text{Mn}^{2+}$  are higher. Fitting the binding data to eq 1 yields the values for  $K_{n1}$ ,  $K_{n2}$ , and  $K_{n\text{in}}$  shown in Table 1. Shown in Figure 2 are sample data for variants at the same positions as those shown in Figure 1. Each of the sites is apparently competed for by  $\text{Mg}^{2+}$  as shown by the increase in all three apparent dissociation constants for WT-PPase as a function of added  $\text{Mg}^{2+}$ . Paralleling the results with  $\text{Mg}^{2+}$ , most of the variants tested showed only minor effects on  $K_{n1}$ ,  $K_{n2}$ , and  $K_{n\text{in}}$ , and D65E substitution gave a large (14-fold) increase in  $K_{n1}$ . Three notable differences between  $\text{Mn}^{2+}$  and  $\text{Mg}^{2+}$  binding are that D102E substitution has little effect on  $K_{n1}$ , D65E substitution has little effect on  $K_{n\text{in}}$ , and R43K substitution results in a 1.8-fold decrease in  $K_{n2}$ .

**$\text{Mg}^{2+}$  Binding in the Presence of  $P_i$ .** Measured at pH 7.2, added  $P_i$  (20 mM) raises the stoichiometry of  $\text{Mg}^{2+}$  bound to WT-PPase (Figure 3), corresponding to the formation of the  $\text{Mg}_2\text{E}(\text{MgP}_i)_2$  complex, in equilibrium with  $\text{Mg}_2\text{EMg}_2\text{-PP}_i$  (Scheme 1). The data obtained for WT- and variant PPases were fit to eq 2, as described under Calculations, giving the parameter values shown in Table 2. Comparing Tables 1 and 2, it is clear that for all of the PPases tested

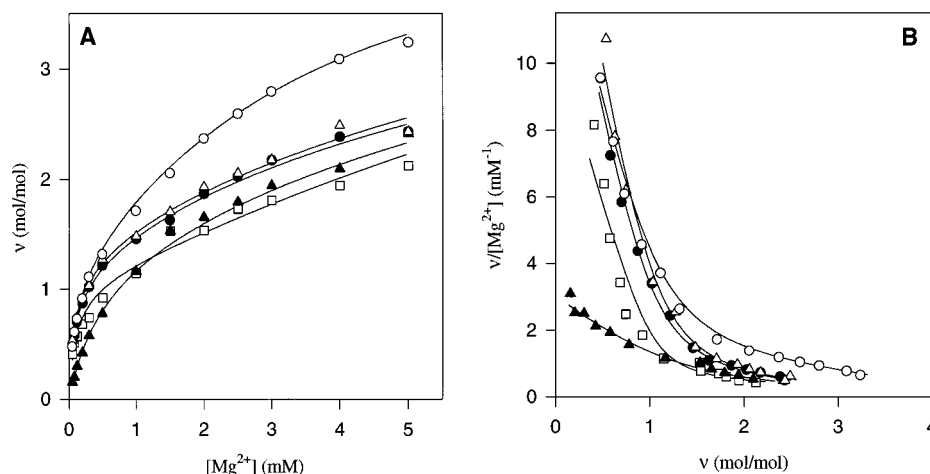


FIGURE 3:  $\text{Mg}^{2+}$  binding to WT-E-PPase and several E-PPase variants in the presence of 20 mM  $P_i$ . Lines are fit to eq 2, using parameter values found in Table 2. Wild-type ( $\circ$ ), E20D ( $\square$ ), K29R ( $\bullet$ ), D65E ( $\blacktriangle$ ), K142R ( $\triangle$ ). (A) Binding curves. (B) Scatchard plots. Data are shown for variants having weaker  $\text{Mg}^{2+}$  binding than wild-type, i.e., increased  $R^{\text{pov}}$ , (see Table 2).



Table 2:  $\text{Mg}^{2+}$  Binding in the Presence of 20 mM  $\text{P}_i$ <sup>a</sup>

E-PPase	$K_{g1}^p$ (mM)	$K_{g2}^p$ (mM)	$K_{g3/4}^p$ (mM)	$R_{ov}^p$
WT	0.069 ± 0.003	1.1 ± 0.1	4.3 ± 0.3	1.00 ± 0.13
E20D	0.093 ± 0.007	4.5 ± 1.2	10 ± 3	20 ± 10
K29R	0.066 ± 0.003	1.9 ± 0.2	12 ± 1.5	12 ± 2
E31D	0.068 ± 0.002	1.9 ± 0.2	10 ± 1	9 ± 2
R43K	0.064 ± 0.002	1.7 ± 0.2	7.9 ± 0.6	5 ± 1
Y55F	0.056 ± 0.002	1.2 ± 0.1	10 ± 1	5 ± 1
D65E	0.34 ± 0.01	2.3 ± 0.3	12 ± 2	14 ± 4
D67E	0.082 ± 0.004	1.4 ± 0.2	4.8 ± 0.4	1.3 ± 0.2
D70E	0.073 ± 0.003	1.9 ± 0.2	12 ± 2	12 ± 3
D97E	0.061 ± 0.003	1.3 ± 0.2	4.2 ± 0.3	1.1 ± 0.2
D102E	0.20 ± 0.01	0.66 ± 0.07	5.9 ± 0.5	1.2 ± 0.2
K104R	0.065 ± 0.004	1.5 ± 0.2	6.2 ± 0.6	2.7 ± 0.5
Y141F	0.051 ± 0.003	2.8 ± 0.1	2.8 ± 0.1	1.0 ± 0.1
K142R	0.058 ± 0.003	2.7 ± 0.3	11 ± 1.5	15 ± 3
D97V	0.096 ± 0.006	1.5 ± 0.3	25 ± 12	43 ± 30

<sup>a</sup> Enzyme concentration 0.8–1.0 mM.

$K_{g1}^p$  is approximately equal to  $K_{g1}$ , so that added  $\text{P}_i$  has little effect on binding to site M1.

Adding  $\text{P}_i$  increases the affinity of the second  $\text{Mg}^{2+}$  bound to the active site (i.e.,  $K_{g2} > K_{g2}^p$ ), with the effect on WT-PPase being 6-fold, and the effects on the variants varying from 1.2-fold (E31D) to >11-fold (E20D). As with  $K_{g2}$ , most substitutions have only a modest effect on  $K_{g2}^p$ , and, again, the largest reduction in affinity (4-fold) results from E20D substitution. These similarities suggest that, for WT-PPase and some variants,  $K_{g2}^p$ , like  $K_{g2}$ , provides an approximate measure of dissociation from site M2. However, for other variants  $K_{g2}^p$  is not much less than  $K_{g3/4}^p$ , so that the second, third, and fourth  $\text{Mg}^{2+}$  bound per monomer (exclusive of the interface) would be expected to distribute among sites M2, M3, and M4, eliminating any clear linkage between the macroscopic constants and constants reflecting dissociation from single subsites. Thus, changes in the value of  $K_{g3/4}^p$  as a function of active site substitution reflect changes in binding not only to subsites M3 and M4 but, for some variants, also to subsite M2 as well.

Because variant effects on the values of  $K_{g2}^p$  and  $K_{g3/4}^p$  are unreliable as measures of effects on subsites M2, M3, and M4, we instead define a parameter  $R_{ov}^p$ , equal to the ratio  $[K_{g2}^p(K_{g3/4}^p)^2(\text{variant})/K_{g2}^p(K_{g3/4}^p)^2(\text{wild-type})]$ , to compare variant effects on overall  $\text{Mg}^{2+}$  binding to these subsites. The nonconservative active site variant D97V has the largest  $R_{ov}^p$ , 43. For the conservative variants, large (12–20) values are observed for E20D, K29R, D65E, D70E, and K142R, more moderate values (5–9) for E31D, R43K, and Y55F, and small  $R_{ov}^p$  values (1–3) for D67E, D97E, D102E, K104R, and Y141F.

The utility of comparing  $R_{ov}^p$  values is subject to limitations imposed by the equilibrium dialysis measurement technique and the multiple  $\text{Mg}^{2+}$  sites per subunit. Equilibrium dialysis measures  $\text{Mg}^{2+}$  bound as the difference between total  $\text{Mg}^{2+}$  concentration (the sum of free and bound  $\text{Mg}^{2+}$ ) and free  $\text{Mg}^{2+}$  concentration. This restricts the useful range of free  $\text{Mg}^{2+}$  concentration to 5–10-fold the E-PPase concentration (the highest level employed was 1.0 mM, 20 mg/mL), since beyond this value the error in the difference does not permit accurate values to be obtained. This problem is exacerbated in the present case by the presence of multiple sites: i.e., estimation of a dissociation constant for a single weak site involves determining values of bound  $\text{Mg}^{2+}$ /PPase monomer of between 0.0 and 1.0, whereas estimation of a

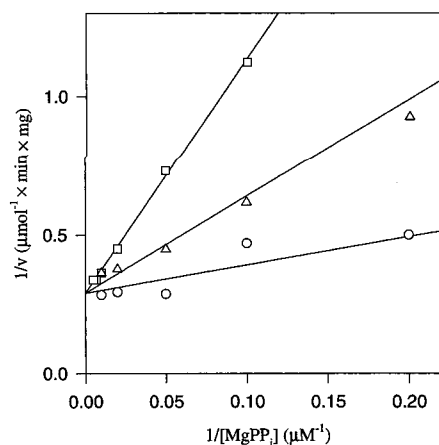


FIGURE 4: HMBP inhibition of the initial rate of WT-E-PPase-catalyzed  $\text{PP}_i$  hydrolysis at 20 mM  $\text{Mg}^{2+}$ . Lines drawn are to eq 4, at  $[\text{HMBP}]$  concentrations ( $\mu\text{M}$ ) of 0 ( $\circ$ ), 80 ( $\Delta$ ), and 240 ( $\square$ ) with  $K_m$  equal to 3.5  $\mu\text{M}$  (30), and  $K_i$  equal to 33  $\mu\text{M}$ .

dissociation constant to a weaker site in the presence of tighter sites requires estimation of such values, for example, between 1.5 and 2.5, or between 2.5 and 3.5, which are inherently less accurate. Restriction of the useful range of free  $\text{Mg}^{2+}$  concentration has as a consequence that sites with dissociation constant values of  $>3$  mM will only be partially occupied, making it difficult to distinguish average effects at each of the subsites M2–M4 from small effects at some and large effects on others. A case in point is the R43K variant. Recent X-ray diffraction results on crystals of the corresponding R78K variant of Y-PPase grown in the presence of  $\text{Mn}^{2+}$  and  $\text{P}_i$  show that while the M1, M2, M4, and P2 sites are filled, the M3 and P1 sites are unoccupied (34). In this case, a moderate  $R_{ov}^p$  value masks a dramatic effect on the M3 subsite.

**$\text{Mg}^{2+}$  Binding in the Presence of HMBP.** HMBP, previously shown to be a competitive inhibitor of yeast (35) and mammalian PPases (36, 37), is also a competitive inhibitor of E-PPase, with a  $K_i$  vs  $\text{PP}_i$  of 33  $\mu\text{M}$ , measured at pH 7.2 and 20 mM  $\text{Mg}^{2+}$  (Figure 4). As seen in Figure 5, added HMBP (100  $\mu\text{M}$ ) parallels the effect of 20 mM  $\text{P}_i$  in inducing binding of additional  $\text{Mg}^{2+}$  per E-PPase subunit, presumably as the complex  $\text{Mg}_2\text{HMBP}$ , by analogy to the binding of  $\text{Mg}_2\text{PP}_i$  in Scheme 1. Proceeding as above, binding results for WT-E-PPase and E-PPase variants were fit to eq 2, yielding the values for  $K_{g1}^p$ ,  $K_{g2}^p$ , and  $K_{g3/4}^p$  presented in Table 3. Also given in Table 3 are values for  $R_{ov}^p$ , equal to the ratio  $[K_{g2}^p(K_{g3/4}^p)^2(\text{variant})/K_{g2}^p(K_{g3/4}^p)^2(\text{wild-type})]$ .

These values show some interesting similarities and differences vis-à-vis the corresponding constants determined in the presence of  $\text{P}_i$ . First, for WT-PPase, the values of these constants are more tightly bunched than the corresponding values measured in the presence of  $\text{P}_i$  [ $K_{g3/4}^p/K_{g1}^p$  (equal to 19)  $<$   $K_{g3/4}^p/K_{g1}^p$  (equal to 62)], resulting from a 2-fold weakening of affinity for the first bound  $\text{Mg}^{2+}$  and a strengthening of third and fourth  $\text{Mg}^{2+}$  binding. Second, D65E substitution results in a large increase (5-fold) in  $K_{g1}^p$ , paralleling the 5–7-fold effects seen on  $K_{g1}$  and  $K_{g1}$ , but the smaller (3-fold) effect that D102E substitution has on these latter constants is not seen on  $K_{g1}^p$ . Third, although there is some correlation between  $R_{ov}^p$  and  $R_{ov}$  values, there are some notable differences: (a) E31D and D65E have much smaller  $R_{ov}^p$  values (0.5, 0.8) than  $R_{ov}$  values (9, 14);

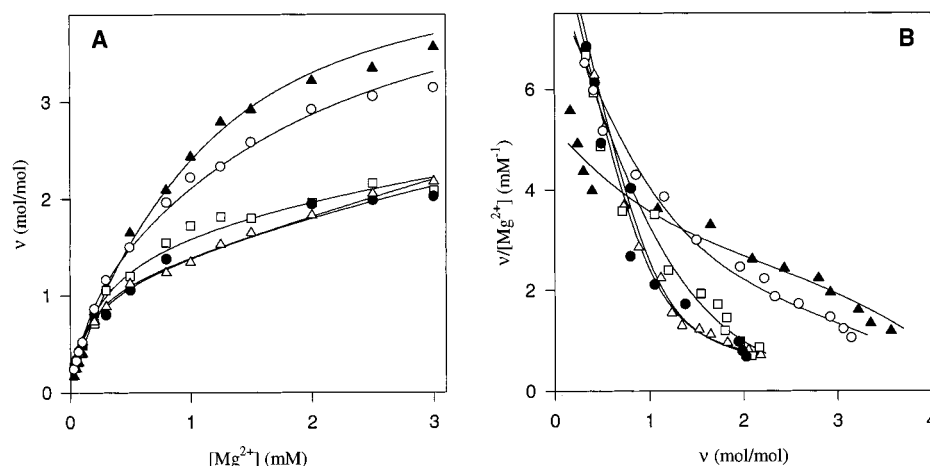


FIGURE 5:  $\text{Mg}^{2+}$  binding to WT-E-PPase and several E-PPase variants in the presence of 100  $\mu\text{M}$  HMBP. Lines are fit to eq 2, using parameter values found in Table 3. Wild-type ( $\circ$ ), E20D ( $\square$ ), D67E ( $\blacktriangle$ ), K104R ( $\bullet$ ), and K142R ( $\triangle$ ). (A) Binding curves. (B) Scatchard plots. Data are shown for three variants having weaker  $\text{Mg}^{2+}$  binding than wild-type, i.e., increased  $R^{\text{pp}}_{\text{ov}}$ , and one variant with stronger  $\text{Mg}^{2+}$  binding than wild-type (see Table 3).

Table 3:  $\text{Mg}^{2+}$  Binding in the Presence of 100 mM HMBP<sup>a</sup>

E-PPase	$K^{\text{pp}}_{\text{g1}}$ (mM)	$K^{\text{pp}}_{\text{g2}}$ (mM)	$K^{\text{pp}}_{\text{g3/4}}$ (mM)	$R^{\text{pp}}_{\text{ov}}$
WT	$0.128 \pm 0.004$	$0.73 \pm 0.07$	$2.4 \pm 0.2$	$1.0 \pm 0.15$
E20D	$0.121 \pm 0.007$	$0.90 \pm 0.15$	$10 \pm 2$	$21 \pm 7$
K29R	$0.119 \pm 0.003$	$1.6 \pm 0.2$	$3.0 \pm 0.3$	$3.5 \pm 0.7$
E31D	$0.084 \pm 0.005$	$1.2 \pm 0.2$	$1.7 \pm 0.2$	$0.8 \pm 0.2$
R43K	$0.108 \pm 0.006$	$3.6 \pm 1.0$	$4.5 \pm 1.3$	$17 \pm 8$
Y55F	$0.088 \pm 0.003$	$2.8 \pm 0.5$	$4.4 \pm 0.7$	$13 \pm 4$
D65E <sup>b</sup>	$0.65 \pm 0.06$	$0.65 \pm 0.06$	$1.8 \pm 0.5$	$0.5 \pm 0.2$
D67E	$0.204 \pm 0.009$	$0.75 \pm 0.11$	$1.44 \pm 0.14$	$0.4 \pm 0.1$
D70E	$0.109 \pm 0.005$	$2.7 \pm 0.6$	$5.1 \pm 1.3$	$17 \pm 7$
D97E	$0.073 \pm 0.005$	$0.83 \pm 0.19$	$1.9 \pm 0.3$	$0.7 \pm 0.2$
D102E <sup>b</sup>	$0.173 \pm 0.003$	$0.173 \pm 0.003$	$1.8 \pm 0.1$	$0.14 \pm 0.01$
K104R	$0.106 \pm 0.006$	$2.5 \pm 0.8$	$8 \pm 3$	$38 \pm 24$
Y141F <sup>c</sup>	$0.097 \pm 0.004$	$2.9 \pm 0.1$	$2.9 \pm 0.1$	$5.8 \pm 0.4$
K142R <sup>c</sup>	$0.107 \pm 0.004$	$5.2 \pm 0.3$	$5.2 \pm 0.3$	$33 \pm 3$
D97V	$0.223 \pm 0.009$	$4.5 \pm 1.3$	$>20$	$>400$

<sup>a</sup> Enzyme concentration 0.8–1.0 mM. <sup>b</sup>  $K^{\text{pp}}_{\text{g1}}$  set equal to  $K^{\text{pp}}_{\text{g2}}$  because the initial fitted values were indistinguishable from one another.

<sup>c</sup>  $K^{\text{pp}}_{\text{g2}}$  set equal to  $K^{\text{pp}}_{\text{g3/4}}$  because the initial fitted values were indistinguishable from one another.

(b) inversely, K104R and Y141F have much larger  $R^{\text{pp}}_{\text{ov}}$  values (38 and 6, respectively) than  $R^{\text{p}}_{\text{ov}}$  values (3 and 1, respectively); and (c) D102E actually enhances average  $\text{Mg}^{2+}$  binding to sites M2–M4 ( $R^{\text{pp}}_{\text{ov}}$  equal to 0.14), but has little effect on  $R^{\text{p}}_{\text{ov}}$ .

**$\text{P}_i$  Binding.** In earlier studies on Y-PPase, we showed there to be a high-affinity  $\text{P}_i$  site ( $K_d \sim 0.2$  mM) and a lower affinity site ( $K_d \sim 1.6$  mM) in the presence of  $\text{Mn}^{2+}$  (38). As shown in Figure 6, in the presence of 100  $\mu\text{M}$   $\text{Mn}^{2+}$ ,  $\text{P}_i$  also binds to a high-affinity site on E-PPase, with a dissociation constant ( $K_p$ ) of 0.16 mM and a stoichiometry of 1.0/subunit. Binding to this site was also examined for the 13 conservative active site variants examined by our previous assays, and for 2 additional variants, K34R and E98V, as well. Most of these variants display decreased binding affinity (Table 4). The most potent effects ( $>50$ -fold increase in  $K_p$ ) are seen for D70E and Y55F substitution. A second group of substitutions, showing increases in  $K_p$  from 8- to 36-fold, include E20D, K29R, E31D, R43K, D65E, and K142R. The remaining substitutions have only minor effects (relative  $K_p$  values of 0.5–3.3) except for D97E, which decreases  $K_p$  9-fold.

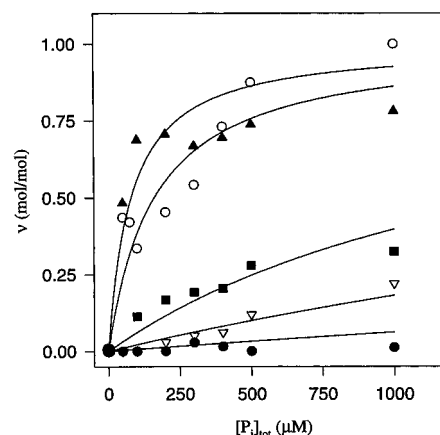


FIGURE 6:  $\text{P}_i$  binding to WT-E-PPase and several E-PPase variants in the presence of 100  $\mu\text{M}$   $\text{Mn}^{2+}$ . Lines are fit to eq 3, using parameter values found in Table 4. Wild-type ( $\circ$ ), K29R ( $\nabla$ ), R43K ( $\blacksquare$ ), D70E ( $\bullet$ ), and D97E ( $\blacktriangle$ ). Data are shown for variants having the full range of effects on  $\text{P}_i$  binding, from enhanced vs wild-type to strongly diminished (see Table 4).

Table 4:  $\text{P}_i$  Binding in the Presence of 100 mM  $\text{Mn}^{2+}$ <sup>a</sup>

E-PPase	$K_p$ (mM)	relative $K_p$
WT	$0.16 \pm 0.02$	1.00
E20D	$1.3 \pm 0.2$	8.1
K29R	$5.0 \pm 0.6$	31
E31D	$5.8 \pm 1.3$	36
R43K	$1.4 \pm 0.1$	8.8
Y55F	$8.7 \pm 2.1$	54
D65E	$3.0 \pm 0.4$	19
D67E	$0.08 \pm 0.01$	0.5
D70E	$>15$	$>90$
D97E	$0.018 \pm 0.004$	0.11
D102E	$0.44 \pm 0.03$	2.8
K104R	$0.40 \pm 0.05$	2.5
Y141F	$0.52 \pm 0.04$	3.3
K142R	$2.4 \pm 0.4$	15
K34R	$0.21 \pm 0.04$	1.3
E98V	$0.51 \pm 0.03$	3.2

<sup>a</sup> Enzyme concentration 0.5 mM.

## DISCUSSION

The experiments reported in this paper are part of a broad effort to fully characterize the structure and function of

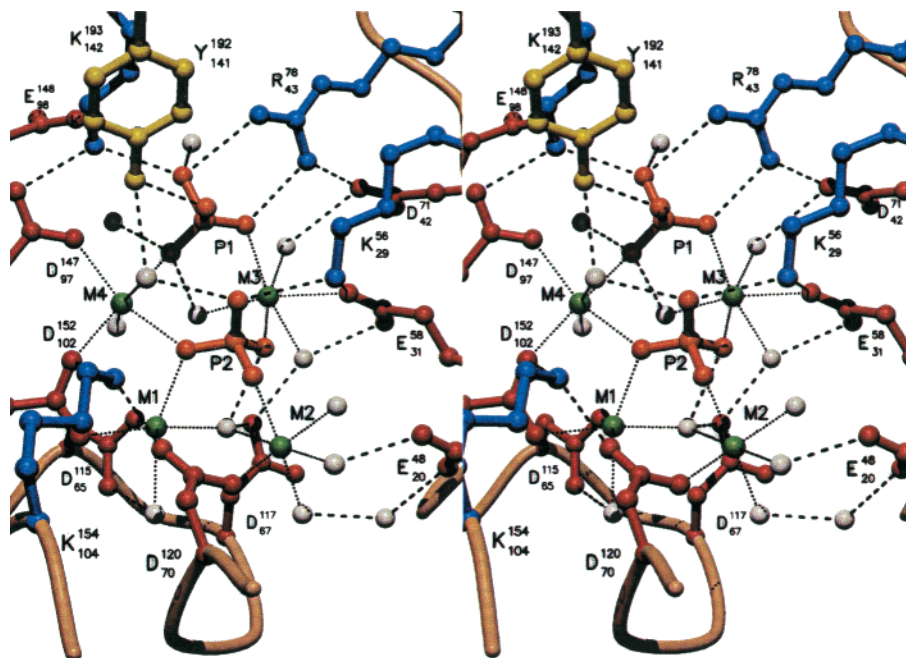


FIGURE 7: Stereo drawing of the active site of the Y-PPase  $\text{Mn}_2\text{E}(\text{MnP}_1)_2$  complex. Acidic residues are in red and basic in blue; tyrosine is in yellow.  $\text{Mn}^{2+}$  ions (M1–M4) are shown in green; phosphates (P1, P2) in orange; and waters in gray. Both E-PPase (lower numbers) and Y-PPase (upper numbers) numbering is shown for the active site residues. Adapted from ref 21.

type I soluble PPases. Here we demonstrate that the sensitivity to conservative substitution of active site residues of ligand binding to E-PPase varies considerably from residue to residue. The present results, considered in light of pertinent results from crystallographic studies on both E-PPase and Y-PPase and functional studies on Y-PPase, allow conclusions regarding (a) substitutions that induce general perturbation of active site structure; (b) specific subsite properties such as functional interaction with other subsites and flexibility; and (c) differences in the structure of enzyme-bound substrate ( $\text{PP}_i$ ) and product ( $2\text{P}_i$ ) complexes.

Interactions at the active site defined by X-ray crystallography (Figure 7) are summarized in Table 5, which includes results for the  $\text{Mg}_{1.5}\text{E}$  (23),  $\text{Mg}_{2.5}\text{E}$  (24), and sulfate (25) complexes of E-PPase, and the  $\text{Mn}_2\text{E}$ ,  $\text{Mn}_2\text{E}(\text{MnP}_1)_2$  (product), and  $\text{Mn}_2\text{EMn}_2\text{FPP}_i$  complexes of Y-PPase (21, 22, 26). These interactions provide an excellent framework for interpreting the effects of active site substitution determined in this work. That the interpretation of the E-PPase results presented in this paper is based in part on Y-PPase structures is reasonable based on the high degree of structural and functional similarity between these two enzymes (15, 20, 21, 39, 40). In what follows we use E-PPase numbering for both E-PPase residues and the aligned, conserved active site residues in Y-PPase (15, 20).

**Subsites M1 and M2 in E-PPase.** Our results for these subsites are reasonably straightforward to interpret because  $K_{g1}$ ,  $K_{n1}$ ,  $K_{p_{g1}}$ , and  $K^{pp_{g1}}$  provide direct measures of metal ion dissociation from M1, and  $K_{g2}$ ,  $K_{n2}$ , and, for the most part,  $K_{p_{g2}}$  provide direct measures of metal ion dissociation from M2.

Particularly striking for M1 is the similarity of the  $K_{g1}$ ,  $K_{p_{g1}}$ , and  $K^{pp_{g1}}$  values, not only for WT-E-PPase but also for each of the conservative active site variants (Tables 1–3). This suggests that site M1 is functionally rather isolated from the remainder of the active site, in the sense that  $\text{P}_i$  or HMBP

Table 5: Correlation of Ligand Contacts<sup>a</sup> and Substitution Effects in PPase

residue or $\text{P}_i$ site <sup>b</sup>	M1	M2	M3	M4	P1	P2	substitution effects <sup>c</sup>
E20(D)	—	vw	—	—	—	—	L/L/M
K29(R)	—	—	—	—	(d) <sup>d</sup>	d	L/M/L
E31(D)	—	—	d	—	—	—	M/S/L
R43(K)	—	—	—	—	d	—	M/L/M
Y55(F)	—	—	—	—	—	d	L/M/VL
D65(E)	d	—	—	vw	—	—	L/S/L
D67(E)	vw	vw	vw	—	—	vw	S/S/S
D70(E)	d	d	—	—	—	—	L/L/VL
D97(E)	—	—	—	d	—	—	S/S/dec
D102(E)	d	—	—	d	—	—	S/dec/S
K104(R)	—	—	—	vw	—	vw	S/L/S
Y141(F)	—	—	—	vw	d	vw	S/M/S
K142(R)	—	—	—	—	d	—	L/L/L
P1	—	—	d	d	—	—	
P2	d	d	d	d	—	—	

<sup>a</sup> As reported in refs 21, 23–25. Abbreviations: d, direct; vw, via water; —, no contact. <sup>b</sup> Parentheses show substitutions, the effects of which are presented in the last column. <sup>c</sup> Effects of substitution on  $R_{p_{ov}}^{pp_{ov}}$ /relative  $K_p$ : VL, >50-fold; L, >10–40-fold; M, 3.5–9-fold; S, 0.4–3-fold; dec, decrease. <sup>d</sup> Seen only in the sulfate complex of E-PPase (25).

binding has little effect on M1 binding. Evidence that  $\text{P}_i$  binds to site P1 in the absence of divalent metal ion comes from the crystal structure of the sulfate complex of E-PPase (25) and from  $\text{P}_i$  inhibition of R43 modification (38). The estimated dissociation constant for  $\text{P}_i$  binding to form  $\text{EP}_i$  is  $\sim 1$  mM (38). Thus,  $K_{p_{g1}}$ , determined in the presence of 20 mM  $\text{P}_i$ , measures  $\text{Mg}^{2+}$  binding to a mixture of E and  $\text{EP}_i$ , in which  $\text{EP}_i$  predominates. In a parallel fashion, the similarity of  $K_{g1}$  and  $K^{pp_{g1}}$  values shows that  $\text{Mg}^{2+}$  affinity is little affected by HMBP at the active site, making the reasonable assumption that this ligand also binds to the active site in the absence of divalent metal ion.

Site M1 interacts directly with D65, D70, and D102, and via water with D67 (Table 5, Figure 7), yet conservative



D/E substitution at only D65 results in a large increase in  $K_{g1}$ ,  $K_{n1}$ ,  $K_{p_{g1}}$ , and  $K_{pp_{g1}}$  values (5–15-fold). The corresponding range for D102E substitution is 1–3-fold, and no other conservative substitution, including D67E and D70E, leads to a >2-fold increase in any of these four constants. In addition, the nonconservative D102V substitution increases  $K_{g1}$  11-fold, placing an upper limit on the contribution of the negative charge on D102 to M1 binding, as contrasted with <2-fold increases on D97V or E98V substitution.

D65E substitution significantly reduces ligand affinity, not only to site M1, but also to other sites, as measured by  $R_{p_{ov}}$ ,  $K_{g_{in}}$ , and  $K_p$ , so that it appears that the structural perturbation resulting from this substitution is propagated elsewhere in the active site as well as to the trimer–trimer interface. On the other hand, D67E and D102E substitutions apparently do not cause long-range perturbations of the active site, since they neither cause major reduction in M1 affinity nor significantly reduce any other measures of ligand affinity reported here (Tables 1–4).

In contrast, D70E substitution, while little affecting binding to M1 or M2, clearly results in major perturbation of the active site. D70E-PPase has by far the lowest enzymatic activity of all of the conservative active site variants we have characterized (16, 20). In addition, despite the lack of direct interaction of D70 with P1, P2, M3, or M4, D70E substitution dramatically decreases not only  $P_i$  binding (Table 4) but also both  $P_i$ - and HMBP-induced  $Mg^{2+}$  binding (Tables 2 and 3). Furthermore, the structure of the related D70N variant of E-PPase shows large-scale reorganization of the active site, despite relatively little motion at N70 with respect to WT-E-PPase (41). It would thus appear that the D70 side chain occupies a critical location for maintaining the overall structure of the active site. We believe it likely that while local flexibility allows the carboxylate side chain in the D70E variant to fully participate in  $M^{2+}$  binding to both M1 and M2, the resulting need to accommodate the extra methylene of the variant induces a distortion that is propagated throughout the active site.

Site M2 interacts directly with D70 and via water with E20 and D67, but of all the conservative variants studied, only E20D substitution substantially increases either  $K_{g2}$  or  $K_{p_{g2}}$ . In addition, E20D substitution promotes dissociation of E-PPase hexamer into trimers (18) and leads to large increases in  $K_{g_{in}}$ ,  $R_{p_{ov}}$ ,  $R_{pp_{ov}}$ , and  $K_p$ , despite the lack of any direct contact of E20 either with subsites other than M2 (Table 5) or with the trimer–trimer interface. Accordingly, it appears not only that E20D substitution distorts the M2 subsite but also that this distortion is propagated throughout the enzyme structure, at the active site and reaching to the trimer–trimer interface.

**Subsites M1 and M2: E-PPase vs Y-PPase.** Despite the extensive structural and functional similarity between the active sites of E-PPase and Y-PPase (20, 21, 40), some interesting differences are also apparent. For example, WT-Y-PPase binds metal ions to sites M1 and M2 37- and 190-fold more tightly than WT-E-PPase, respectively (12). In addition, the effects of mutation of aligned residues in Y-PPase and E-PPase differ both qualitatively and quantitatively. In contrast to what is seen for E-PPase variants (Table 1), every Y-PPase variant examined showed a significant effect on metal ion binding to M2 (42). Further, the sensitivity of M1 binding to two D/E mutations is

inverted. Thus, for E-PPase, D65E substitution significantly decreases affinity and D70E substitution has little effect (Table 1), whereas for Y-PPase D115E substitution has no effect on M1 binding but D120E substitution has a significant effect (42). These differences in metal ion binding properties are most probably due to the closer packing of the Y-PPase active site.

**Subsites M3, M4, P1, and P2.** Addition of either  $P_i$  or HMBP induces binding to M3 and M4. Many active site substitutions affect both  $P_i$ - and HMBP-induced  $Mg^{2+}$  binding (Tables 2 and 3) as well as  $P_i$  binding (Table 4). This is not unexpected, since in all three cases such effects can arise from perturbations at metal-ion and/or phosphoryl group binding subsites. Interpretation of effects on  $K_p$  is further complicated by two additional factors. First, the apparent effects on  $K_p$  may only represent lower limits, since severe distortion of the high-affinity site may result in binding of the first  $P_i$  to the lower affinity site. Second, there is some ambiguity concerning the identity of the high-affinity  $P_i$  site. Zyranov et al. (43) have recently demonstrated for Y-PPase that this identity is divalent metal ion dependent, with P2 being the high-affinity site in the presence of  $Mn^{2+}$ , whereas P1 is the high-affinity site either in the presence of  $Mg^{2+}$  (8, 44) or in the absence of divalent metal ion altogether (see above). Our results for substitution effects on  $K_p$  would tend to support this ordering for E-PPase in the presence of  $Mn^{2+}$  as well. Thus, Y55 binds directly to P2 rather than to P1 (Table 5), and, apart from D70E substitution, which results in a general perturbation of the active site, Y55F substitution causes the largest increase in  $K_p$  (Table 4). By contrast, Y141 binds directly to P1 and not to P2, and Y141F substitution has only a modest effect on  $K_p$ . Moreover, substitutions of aligned residues in E-PPase and Y-PPase have similar weakening effects on high-affinity  $P_i$  binding. Thus, for Y-PPase, the effects fall in the order (in E-PPase numbering) Y55F > K29R > R43K, K142R (43), paralleling what is reported in Table 4.

As a result of these factors, substitution effects cannot, in general, be attributed to single subsites, as is possible for M1 and, for most variants, for M2. Rather, it is more appropriate to consider the parameters  $R_{p_{ov}}$ ,  $R_{pp_{ov}}$ , and  $K_p$  as providing different measures of overall ligand affinity to subsites M3, M4, P1, and P2, and, for some variants, to M2. The effects of conservative substitution on these parameters, as summarized in Table 5, reflect alterations at one or more of these subsites, arising from direct perturbation of a binding interaction, or from a perturbation that is propagated within the active site cavity.

Despite these inherent ambiguities, the results in Table 5 permit three further conclusions: first, that K142, like E20, D65, and D70, plays a key role in the functional organization of the active site, since K142R substitution induces large increases in  $R_{p_{ov}}$ ,  $R_{pp_{ov}}$ , and  $K_p$  as well as in  $K_m$  for  $Mg_2PP_i$  (20); second, that the M4 subsite is an especially unconstrained part of the active site. This conclusion derives from the observation that conservative substitutions of three of the four amino acid side chains interacting directly or via water with this subsite (D97, D102, and Y141) result in no large decreases in ligand affinity (the exception is D65E substitution, which causes a general perturbation). Supporting this conclusion, the D97–D102 loop, including the two residues binding directly to M4, is disordered in some



E-PPase structures (45) and undergoes the largest motions between the Mn<sub>2</sub>-Y-PPase and Mn<sub>2</sub>-Y-PPase(MnPi)<sub>2</sub> structures (21). The third conclusion is that the major differences between the product and substrate complexes of E-PPase are concentrated in the lower half of the active site (Figure 7). This conclusion, which is consistent with recent structural studies on Y-PPase (22), is based on the observation that the four substitutions showing the greatest differences in effects on  $R_{\text{P}_{\text{ov}}}$  vs  $R_{\text{P}_{\text{ov}}}$ , E31D, D65E, D102E, and K104R, all fall within the lower half of the active site. Moreover, the polar termini of D65, D102, and K104 are quite close to one another (Figure 7 and ref 16) and are centered about D102, within the flexible 97–102 loop.

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